

# Three-Dimensional Surface Treatment of MoS<sub>2</sub> Using BCl<sub>3</sub> Plasma-Derived Radicals

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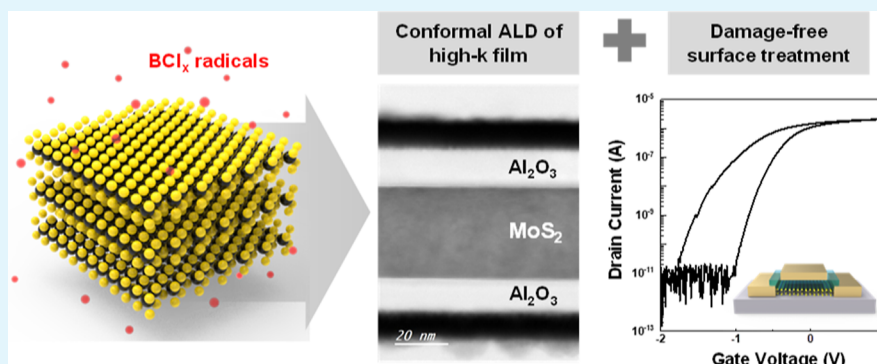
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**ABSTRACT:** The realization of next-generation gate-all-around field-effect transistors (FETs) using two-dimensional transition metal dichalcogenide (TMDC) semiconductors necessitates the exploration of a three-dimensional (3D) and damage-free surface treatment method to achieve uniform atomic layer-deposition (ALD) of a high-k dielectric film on the inert surface of a TMDC channel. This study developed a BCl<sub>3</sub> plasma-derived radical treatment for MoS<sub>2</sub> to functionalize MoS<sub>2</sub> surfaces for the subsequent ALD of an ultrathin Al<sub>2</sub>O<sub>3</sub> film. Microstructural verification demonstrated a complete coverage of an approximately 2 nm-thick Al<sub>2</sub>O<sub>3</sub> film on a planar MoS<sub>2</sub> surface, and the applicability of the technique to 3D structures was confirmed using a suspended MoS<sub>2</sub> channel floating from the substrate. Density functional theory calculations supported by optical emission spectroscopy and X-ray photoelectron spectroscopy measurements revealed that BCl radicals, predominantly generated by the BCl<sub>3</sub> plasma, adsorbed on MoS<sub>2</sub> and facilitated the uniform nucleation of ultrathin ALD–Al<sub>2</sub>O<sub>3</sub> films. Raman and photoluminescence measurements of monolayer MoS<sub>2</sub> and electrical measurements of a bottom-gated FET confirmed negligible damage caused by the BCl<sub>3</sub> plasma-derived radical treatment. Finally, the successful operation of a top-gated FET with an ultrathin ALD–Al<sub>2</sub>O<sub>3</sub> (~5 nm) gate dielectric film was demonstrated, indicating the effectiveness of the pretreatment.

**KEYWORDS:** MoS<sub>2</sub>, surface functionalization, BCl<sub>3</sub> plasma, radicals, atomic layer deposition, high-k dielectric

## 1. INTRODUCTION

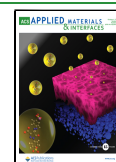
For a continuous technical node scaling of Si field-effect transistors (FETs), gate-all-around FETs (GAAFETs) with multiple nanosheet- or nanowire-shaped channels completely wrapped with a high-k gate dielectric film are projected to gradually replace FETs with fin structures.<sup>1,2</sup> Moreover, ultrathin Si channels in GAAFETs are expected to be replaced with alternative semiconductors in the future, such as transition metal dichalcogenides (TMDCs) with two-dimensional (2D) layer structures, which are more immune to mobility degradation and short-channel effects than Si, even at subnanometer thicknesses.<sup>3,4</sup> Thus, to ensure uniform encapsulation of TMDC channel layers within a three-dimensional (3D) GAAFET structure, it is essential to employ an atomic layer deposition (ALD) process that allows the deposition of an ultrathin high-k gate dielectric film with

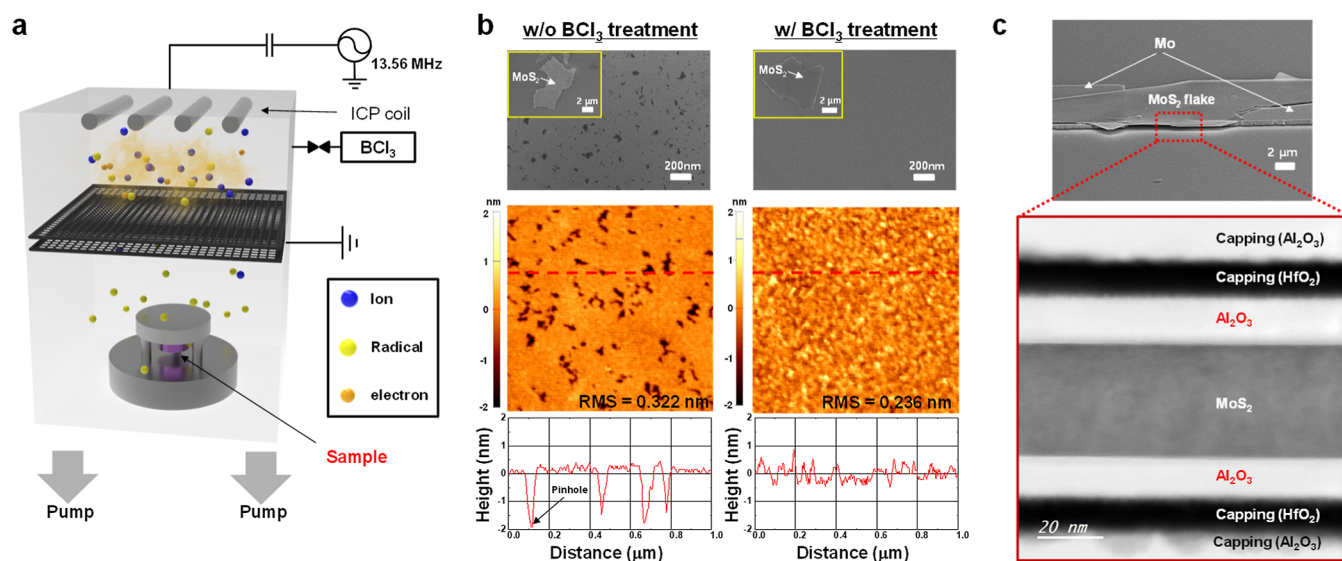
excellent step coverage capabilities and electrical properties.<sup>5,6</sup> However, the absence of chemically reactive bonds on TMDC surfaces hinders the uniform growth of ultrathin high-k gate dielectric films during ALD at high temperatures, necessitating the development of various surface treatments and passivation techniques.<sup>7</sup> In addition, to implement a GAAFET with multiple TMDC channels, it is necessary to develop a damage-free 3D surface-treatment method.

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**Figure 1.** ALD coverage improvement by BCl<sub>3</sub> plasma-derived radical treatment and its applicability to a 3D structure. (a) Schematic of the remote plasma chamber used for BCl<sub>3</sub> plasma-derived radical treatment. (b) SEM (top) and AFM (middle and bottom) results of ALD–Al<sub>2</sub>O<sub>3</sub> (~2 nm) on MoS<sub>2</sub> flakes with and without BCl<sub>3</sub> plasma-derived radical treatment. (c) TEM analysis results of the ALD of Al<sub>2</sub>O<sub>3</sub> on the BCl<sub>3</sub> plasma-derived radical-treated MoS<sub>2</sub> flakes suspended by patterned Mo supports.

Although many approaches have been proposed to enhance the deposition uniformity and thickness scalability of high-*k* films on 2D TMDC surfaces,<sup>7</sup> little attention has been paid to addressing these issues on 3D TMDC substrates. One potential technique for the surface treatment of 3D structures is the use of chemically reactive and electrically neutral radicals generated in a plasma process, without directionality. Previous studies utilized O<sub>2</sub><sup>8–10</sup> and Cl<sub>2</sub><sup>11</sup> plasmas for the treatment of 2D TMDC surfaces. The O<sub>2</sub> plasma treatment significantly improved the coverage of subsequent ALD high-*k* films on MoS<sub>2</sub>.<sup>8,9</sup> However, as a side effect, the simultaneous formation of Mo oxide may increase the off-current of FETs.<sup>10</sup> Although the Cl<sub>2</sub> plasma treatment improved the MoS<sub>2</sub> FET performance, it was only used to modify the MoS<sub>2</sub> channel surface in a bottom-gated FET configuration without subsequent deposition of high-*k* films.<sup>11</sup>

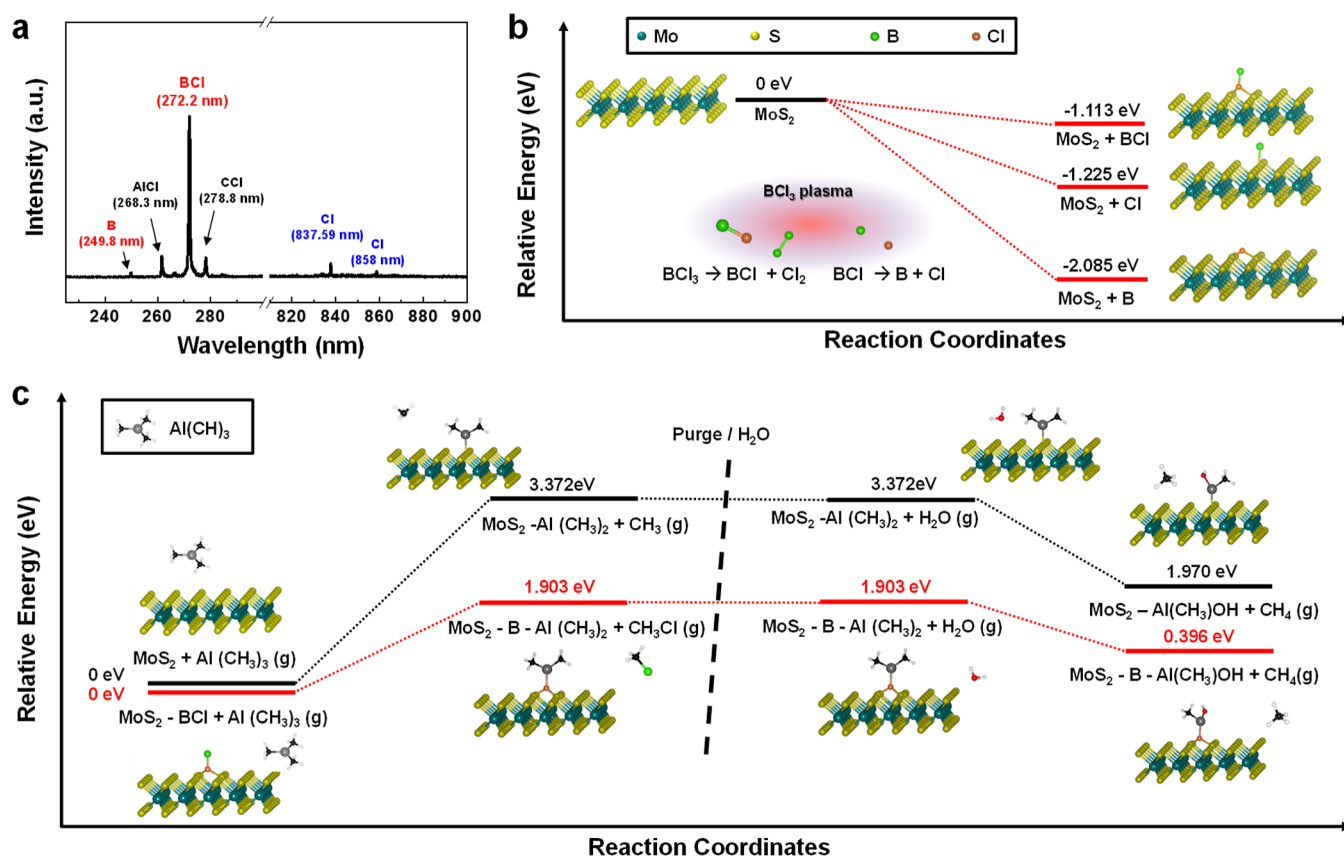
This study used a novel 3D and nondestructive treatment approach for MoS<sub>2</sub> surfaces that employs BCl<sub>3</sub> plasma-derived radicals to promote uniform nucleation of high-*k* gate dielectric films in subsequent ALD.<sup>12</sup> To demonstrate the 3D surface treatment capability of BCl<sub>3</sub>-generated radicals, a remote plasma system with an ion-filtering mesh was utilized and the sample was placed upside down. The facile and uniform growth of an ultrathin Al<sub>2</sub>O<sub>3</sub> film using ALD was demonstrated, which exhibited outstanding compatibility with 3D structures. Additionally, the working mechanism of this process was proposed based on density functional theory (DFT) calculations. Furthermore, the absence of physical damage during treatment was verified by fabricating and electrically characterizing the MoS<sub>2</sub> FETs with bottom- and top-gated structures.

## 2. EXPERIMENTAL DETAILS

**2.1. BCl<sub>3</sub> Plasma-Derived Radical Treatment of MoS<sub>2</sub>.** A 13.56 MHz AC inductively coupled plasma (ICP) generator was used to generate the BCl<sub>3</sub> remote plasma. Figure 1a illustrates the custom-built remote ICP reactor used in this study, which underwent modifications involving the insertion of two electrically grounded mesh grids made of stainless steel with a thickness of 0.14 mm. These

mesh grids were identical, featuring a square array of square holes measuring 0.135 × 0.135 mm. They were overlapped without twisting and were spaced approximately 2 cm apart. The primary purpose of this modification was to use the mesh grids as a plasma shield to prevent the ions produced by the plasma from reaching the substrate, effectively covering an area of 30 × 30 cm, consistent with the chamber dimensions. The use of two mesh grids aimed to enhance the prevention of ions more effectively. In addition, to indirectly demonstrate the potential applicability of the proposed process to 3D-structured devices such as GAAFETs, the specimens were attached to an inverted 4-inch Si wafer that was floated approximately 1.5 mm above the bottom of the plasma chamber. The optimal BCl<sub>3</sub> plasma parameters in our system, such as plasma power (250 W), working pressure (2.1 mTorr), and treatment time (300 s), were preselected by comparing the surface coverage of the ALD–Al<sub>2</sub>O<sub>3</sub> thin films on the single crystalline MoS<sub>2</sub> flakes using plan-view scanning electron microscopy (SEM), as shown in Figure S1. As soon as the treatment was completed, the samples were loaded into the ALD chamber within 5 min to minimize their exposure to ambient air. The chemical species produced by BCl<sub>3</sub> plasma were analyzed through optical emission spectroscopy (OES, Isoplane SCT 3200) using a grating with a blaze wavelength of 300 nm and a groove density of 1200 mm<sup>-1</sup>.

**2.2. ALD of Al<sub>2</sub>O<sub>3</sub> and Fabrication of MoS<sub>2</sub> FETs.** Heavily p-type-doped Si substrates with thick SiO<sub>2</sub> films (90 nm) were cleaned with acetone and isopropyl alcohol (IPA) for 3 min each using a sonicator, and then thick MoS<sub>2</sub> flakes mechanically exfoliated from a bulk crystal (470MOS2M-AB, SPI Supplies) were transferred onto the substrates using conventional 3 M scotch tape. Any remaining tape residues on the MoS<sub>2</sub> flakes were removed by immersing the samples in acetone for 1 h, followed by rinsing with IPA. The ALD of Al<sub>2</sub>O<sub>3</sub> was conducted at a process pressure of ~0.1 Torr in a laboratory-scale and custom-built thermal ALD system with a traveling wave design, using trimethylaluminum (TMA) and H<sub>2</sub>O precursors. Because lower temperatures are known to improve film coverage on MoS<sub>2</sub> by suppressing the thermal desorption of precursors,<sup>13</sup> the lowest deposition temperature of 200 °C was selected while remaining within the typical Al<sub>2</sub>O<sub>3</sub> ALD window of 200–300 °C.<sup>14</sup> Each ALD cycle consisted of a series of injections of TMA (1 s), N<sub>2</sub> (20 s), H<sub>2</sub>O (5 s), and N<sub>2</sub> (30 s). The TMA and H<sub>2</sub>O canisters were maintained at a temperature of 9 °C, and their delivery lines were heated to 60 and 100 °C, respectively. The thickness of the Al<sub>2</sub>O<sub>3</sub> films was monitored on dummy Si substrates cleaned with a 1%



**Figure 2.** Mechanism analyses for the  $\text{BCl}_3$  plasma-derived radical treatment and the subsequent  $\text{Al}_2\text{O}_3$  ALD process. (a) OES analysis result of  $\text{BCl}_3$  plasma. (b) DFT modeling of the adsorption energy reaction pathways of  $\text{BCl}_3$  plasma-generated radicals on the  $\text{MoS}_2$  surface. (c) Comparison of the energy reaction pathways of the ALD- $\text{Al}_2\text{O}_3$  deposition process on  $\text{MoS}_2$  surfaces with and without adsorbed  $\text{BCl}$  radicals, based on DFT modeling.

HF solution, which was subjected to a simultaneous ALD process with the  $\text{MoS}_2$  substrates. Thickness measurements were conducted using a spectroscopic ellipsometer (SE MG-1000UZ, NANO-VIEW) at the MEMS Sensor Platform Center of Sungkyunkwan University. The deposition rate of the  $\text{Al}_2\text{O}_3$  film on Si was approximately 1 Å/cycle at 200 °C.

To fabricate both bottom- and top-gated  $\text{MoS}_2$  FETs, a Cr/Au source/drain layer with a thickness of 10 nm/50 nm was deposited using e-beam evaporation followed by a lift-off patterning process. An identical Cr/Au layer was used for the top-gate electrode in a top-gated  $\text{MoS}_2$  FET after the deposition of the  $\text{Al}_2\text{O}_3$  top-gate dielectric on  $\text{MoS}_2$  pretreated with  $\text{BCl}_3$  plasma-derived radicals.

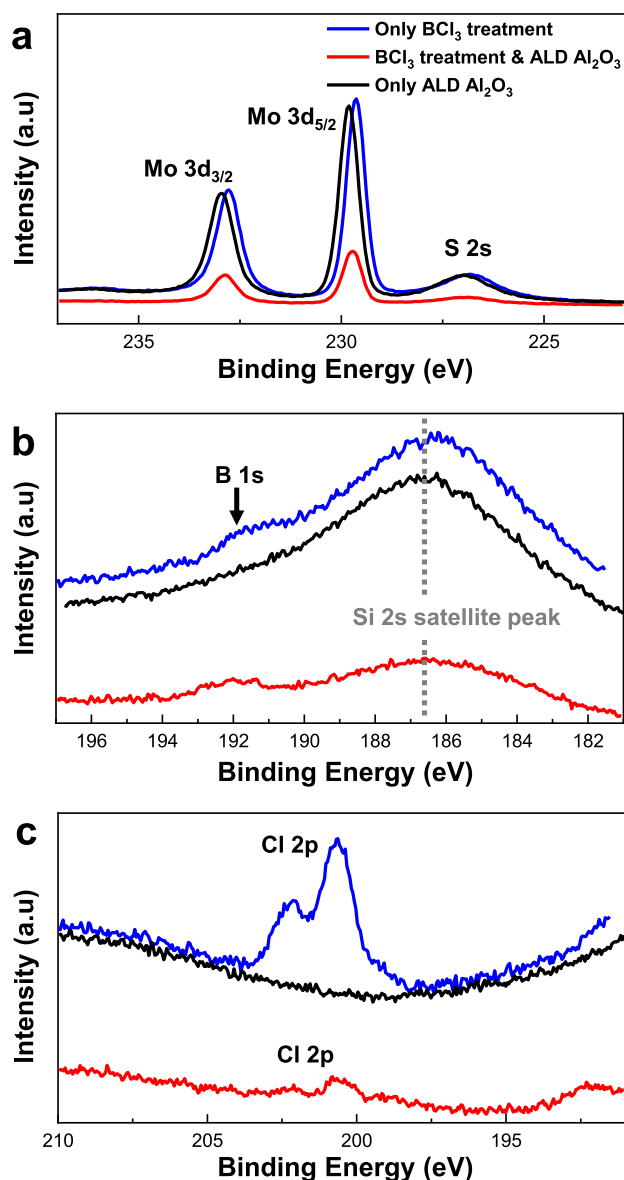
**2.3. Characterization Methods and DFT Calculations.** The surface coverage of the ALD- $\text{Al}_2\text{O}_3$  films on  $\text{MoS}_2$  was examined using SEM (S-4700, Hitachi). In addition, cross-sectional transmission electron microscopy (TEM, ARM 200F, JEOL) was used to observe the conformal deposition characteristics of ALD- $\text{Al}_2\text{O}_3$  on  $\text{MoS}_2$  flakes pretreated with  $\text{BCl}_3$  plasma-derived radicals. The TEM samples were prepared using a focused ion beam (SMI 3050TB, SII NanoTechnology Inc.) system. The possible changes of chemical bonds on the  $\text{MoS}_2$  surface during the  $\text{BCl}_3$  plasma-derived radical treatment and/or ALD- $\text{Al}_2\text{O}_3$  were examined using X-ray photoelectron spectroscopy (XPS, MultiLab 2000, Thermo VG) with an Mg  $K\alpha$  source. A Raman and photoluminescence (PL) spectrometer (Alpha 300M+, Witec) was used to identify and characterize the monolayer  $\text{MoS}_2$  flakes transferred onto  $\text{SiO}_2$  to evaluate the  $\text{BCl}_3$  plasma-derived radical treatment effect. The electrical properties of the fabricated  $\text{MoS}_2$  FETs were characterized using a Keysight B1500A semiconductor parameter analyzer.

To examine the energy changes during the chemical process and the effects of the  $\text{BCl}_3$  plasma-derived radical treatment, DFT calculations were performed using a Vienna ab initio simulation

package (VASP) with a PBEsol functional.<sup>15–17</sup> The first step involved geometric optimization of the  $\text{MoS}_2$  unit cell, which was performed until a convergence condition of 0.01 eV/Å was satisfied using  $11 \times 11 \times 3$   $k$ -points for a  $k$ -spacing of  $<0.2/\text{Å}$  and a cutoff energy of 500 eV. To calculate the formation energy of the  $\text{MoS}_2$ -molecular chemical bonds that could be generated during the ALD process, a  $5 \times 5$  supercell with one layer of  $\text{MoS}_2$  structure was generated from the optimized unit cell.  $\Gamma$  (gamma)  $k$ -points and  $3 \times 3 \times 1$   $k$ -points were used for geometry optimization and formation energy calculations. Based on the difference in the formation energy between the reactant and product, we evaluated the ALD reaction pathway that synthesized  $\text{Al}_2\text{O}_3$  on  $\text{MoS}_2$  with TMA and  $\text{H}_2\text{O}$ .

## 3. RESULTS AND DISCUSSION

**3.1. Coverage Improvement and Applicability to 3D Structures.** First, the improvement in the ALD- $\text{Al}_2\text{O}_3$  film coverage on  $\text{MoS}_2$  through  $\text{BCl}_3$  plasma-derived radical treatment was demonstrated, as shown in Figure 1b. The  $\text{Al}_2\text{O}_3$  films were deposited on both  $\text{BCl}_3$  plasma-treated and untreated thick  $\text{MoS}_2$  flakes for 20 ALD cycles, resulting in the formation of approximately 2 nm-thick  $\text{Al}_2\text{O}_3$  films on Si at 200 °C. Subsequently, the surface coverage of the films was examined using SEM and AFM. On the pristine surface of  $\text{MoS}_2$  without any treatment, ALD led to the formation of  $\text{Al}_2\text{O}_3$  islands with numerous pinholes, indicating incomplete coverage, as observed using SEM and AFM. This phenomenon can be attributed to the higher degree of physical adsorption of the ALD precursors rather than their chemical bonding with  $\text{MoS}_2$ , as reported previously.<sup>7</sup> In contrast, after the  $\text{BCl}_3$  plasma-derived radical treatment, the formation of a



**Figure 3.** Results of the chemical analysis after various surface passivation. (a) Mo 3d and S 2s, (b) B 1s, and (c) Cl 2p XPS peaks observed in the MoS<sub>2</sub> flakes that underwent different surface passivation: only BCl<sub>3</sub> plasma-derived radical treatment (blue line), BCl<sub>3</sub> plasma-derived radical treatment followed by ALD of Al<sub>2</sub>O<sub>3</sub> (red line), and only ALD of Al<sub>2</sub>O<sub>3</sub> (black line).

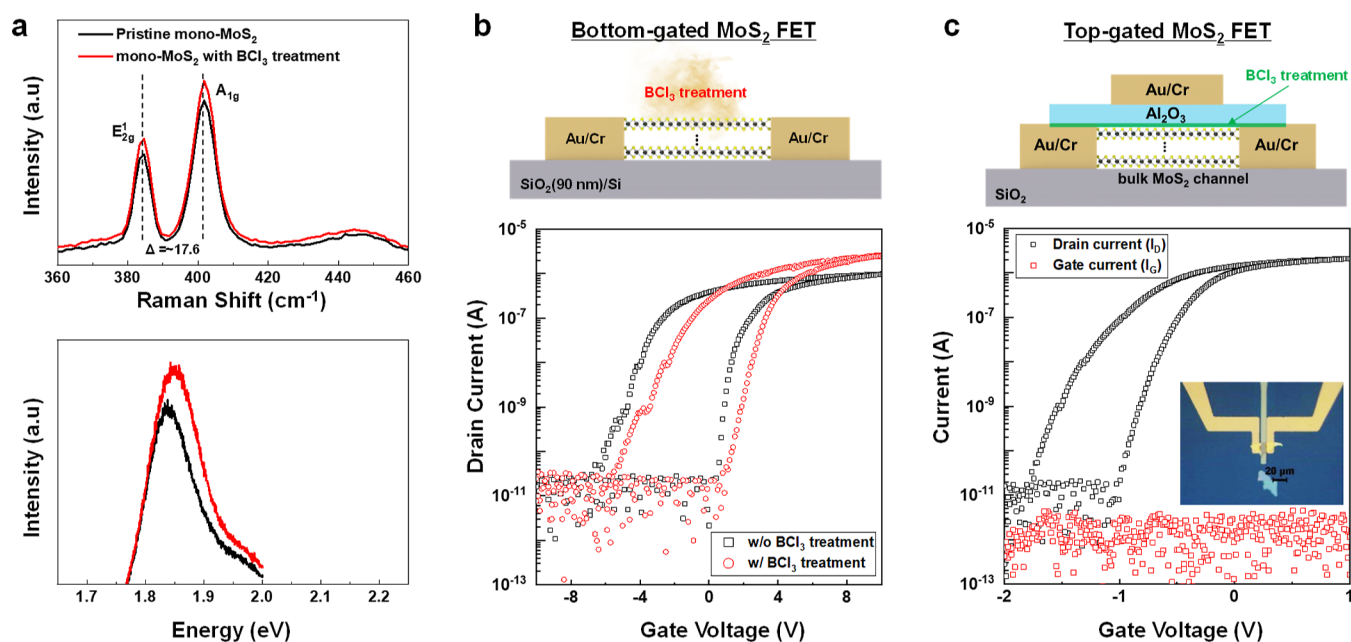
continuous Al<sub>2</sub>O<sub>3</sub> thin film was identified, as confirmed by the corresponding SEM and AFM images (Figure 1b), which exhibited conformal deposition without any detectable pin-holes.

The hardware configuration of the BCl<sub>3</sub> plasma-derived radical treatment system was specifically designed to expose the MoS<sub>2</sub> surface to radical species with minimal directionality (Figure 1a). This feature enables the application of BCl<sub>3</sub> plasma-derived radical treatment to 3D sample structures, facilitating the conformal wrapping of a 2D sheet channel with a high-*k* dielectric film, which is necessary for GAAFET design. To demonstrate the potential applicability of the BCl<sub>3</sub> plasma-derived radical treatment to a 3D-structured device, a MoS<sub>2</sub> bridge structure resembling an actual GAAFET device was fabricated, as shown in Figure 1c. The bridge structure was designed to suspend the thick MoS<sub>2</sub> flakes above the SiO<sub>2</sub>

substrate using two separate Mo (thickness of ~500 nm) supports. These supports were sputter-deposited and patterned using photolithography and etching with a conventional wet etchant (Mo-72, purchased from WINCHEM Co.); the detailed steps of the fabrication are illustrated in Figure S2. After the BCl<sub>3</sub> plasma-derived radical treatment and ALD–Al<sub>2</sub>O<sub>3</sub> film deposition (target thickness of ~12 nm on Si) on the fabricated bridge structure, a cross-sectional TEM sample was prepared and analyzed, as shown in Figure 1c. To provide structural stabilization for the suspended MoS<sub>2</sub> flake during TEM sampling, an additional layer of the ALD–Al<sub>2</sub>O<sub>3</sub> film with a thickness of approximately 200 nm was deposited to fill the gap between the MoS<sub>2</sub> and the substrate. In addition, to differentiate the two Al<sub>2</sub>O<sub>3</sub> layers, an approximately 10 nm-thick ALD–HfO<sub>2</sub> film was inserted between them. Figure 1c shows the conformal deposition of an ALD–Al<sub>2</sub>O<sub>3</sub> film, although a slightly thinner layer was deposited on the bottom side of the MoS<sub>2</sub> flakes, possibly due to an insufficient supply of ALD precursors.

**3.2. Theoretical Mechanism Analyses.** DFT calculations were conducted to investigate the mechanism by which the BCl<sub>3</sub> plasma-derived radical treatment enhanced ALD deposition on MoS<sub>2</sub>. Our models included the adsorption of plasma-generated radicals and ALD precursors, while the atmospheric adsorbates generated during the ex situ process were not taken into consideration. The stable pathway for these reactions was determined by comparing the molecular formation energies before and after the reaction. Before DFT modeling of the plasma process was performed, an OES tool was used to experimentally examine the chemical species produced by the BCl<sub>3</sub> plasma. The spectra of the OES in Figure 2a confirm the generation of a small amount of B and Cl radicals as well as a dominant amount of BCl radicals within the plasma. A small peak corresponding to AlCl observed in the OES spectrum was likely a result of the chemical reaction between BCl<sub>3</sub> and the Al chamber wall. To check the possibility of enhanced nucleation of Al<sub>2</sub>O<sub>3</sub> on MoS<sub>2</sub> through the production of AlCl during BCl<sub>3</sub> plasma pretreatment, a separate experiment was conducted using a similarly configured BCl<sub>3</sub> plasma system with an Al-free chamber. The results showed a similar improvement in the coverage of Al<sub>2</sub>O<sub>3</sub> on MoS<sub>2</sub>, suggesting that the coverage improvement observed in the main experiment was mostly induced by the presence of B, Cl, and BCl radicals rather than by the influence of the Al chamber wall.

The DFT calculation results shown in Figure 2b indicate that the energy reaction of the produced B, Cl, and BCl radicals enabled their stable adsorption onto the MoS<sub>2</sub> surface. This adsorption served as a mediator, facilitating the binding of the metal precursor during the subsequent ALD process. Consequently, the adsorption of TMA molecules onto the preadsorbed B, Cl, and BCl radicals on the MoS<sub>2</sub> surface was found to be a more energetically stable reaction pathway than direct adsorption on the MoS<sub>2</sub> surface. For instance, when the BCl radicals, identified as the predominant species in the plasma according to OES analysis, were preadsorbed on MoS<sub>2</sub> followed by the ALD–Al<sub>2</sub>O<sub>3</sub> deposition process (Figure 2c), the Cl atoms readily transitioned to a gaseous state because of their strong binding to the CH<sub>3</sub> groups of TMA. Consequently, stable MoS<sub>2</sub>–B–Al(CH<sub>3</sub>)<sub>2</sub> bonds were formed as the final products on the MoS<sub>2</sub> surface. Therefore, the ALD pathway with BCl radical treatment was more energetically favorable than direct ALD deposition, and the energy of the



**Figure 4.** Electrical characteristics after  $\text{BCl}_3$  plasma-derived radical treatment. (a) Raman and PL spectra of a monolayer  $\text{MoS}_2$  flake with and without  $\text{BCl}_3$  plasma-derived radical treatment. (b) Schematic (top) and transfer curve (bottom) of a bottom-gated bulk  $\text{MoS}_2$  FET (channel length and width of  $10 \mu\text{m}$  each) before and after  $\text{BCl}_3$  plasma-derived radical treatment. (c) Schematic (top) and transfer curve (bottom) of a top-gated bulk  $\text{MoS}_2$  FET with  $\text{BCl}_3$  plasma-derived radical treatment followed by ALD of  $\text{Al}_2\text{O}_3$  ( $\sim 5 \text{ nm}$ ) as the top gate dielectric. Inset in panel (c) is an optical image of a fabricated top-gated bulk  $\text{MoS}_2$  FET (channel length and width of 10 and  $15 \mu\text{m}$ , respectively).

final state of  $\text{MoS}_2\text{-B-Al}(\text{CH}_3)_2$  was approximately 80% lower than that of  $\text{MoS}_2\text{-Al}(\text{CH}_3)_2$  in the ALD process without any pretreatment.

To validate the aforementioned DFT calculation model experimentally, XPS measurements were conducted on thick  $\text{MoS}_2$  flakes coated with an ALD- $\text{Al}_2\text{O}_3$  ( $\sim 2 \text{ nm}$ ) film, both with and without  $\text{BCl}_3$  plasma pretreatment, as shown in Figure 3. The C 1s peak with a binding energy of 284.8 eV was used as the calibration reference for all the measured XPS spectra.<sup>18</sup> In the pretreated  $\text{MoS}_2$  samples before and after  $\text{Al}_2\text{O}_3$  deposition, B 1s ( $\sim 192 \text{ eV}$ ) and Cl 2p ( $\sim 201$  and  $\sim 200 \text{ eV}$  for  $2p_{1/2}$  and  $2p_{3/2}$  peaks, respectively) peaks were observed. These observations confirmed the facile chemical adsorption of B, Cl, and BCl radicals on  $\text{MoS}_2$ , consistent with the results of the DFT calculations. Notably, the intensity of the Mo peak was significantly reduced following the deposition of  $\text{Al}_2\text{O}_3$  on the  $\text{MoS}_2$  surface pretreated with the  $\text{BCl}_3$  plasma-derived radicals because of the  $\text{Al}_2\text{O}_3$  thickness effect, whereas a negligible change in the Mo peak intensity was observed for the untreated sample. This finding again confirms that the  $\text{BCl}_3$  plasma-derived radical treatment enhanced the nucleation of  $\text{Al}_2\text{O}_3$  and led to enhanced coverage on the  $\text{MoS}_2$  surface.

### 3.3. Possible Physical Damage and FET Performance.

As discussed in the experimental section, the hardware configuration for the  $\text{BCl}_3$  plasma-derived radical treatment was carefully optimized to minimize physical damage to the  $\text{MoS}_2$  surface by removing the ionic species produced by the  $\text{BCl}_3$  plasma. To directly assess the possible adverse effects of the treatment on the  $\text{MoS}_2$  surface, a monolayer  $\text{MoS}_2$  flake prepared using the gold exfoliation method<sup>19</sup> was used for Raman and PL measurements, as shown in Figure 4a. In the Raman spectrum, distinctive peaks associated with  $\text{MoS}_2$ , including the in-plane mode ( $E_{2g}$ ) and the out-of-plane mode ( $A_{1g}$ ), were observed at  $\sim 384.1$  and  $\sim 401.7 \text{ cm}^{-1}$ , respectively, when using a 532 nm excitation laser wavelength. The distance

between these two peaks ( $\sim 17.6 \text{ cm}^{-1}$ ) was used to experimentally confirm the  $\text{MoS}_2$  monolayer following a reference source.<sup>20</sup> Compared with the pristine state, the Raman peak positions of  $\text{MoS}_2$  remained nearly unchanged, and no decrease in their intensity was observed, even after treatment with a monolayer of  $\text{MoS}_2$ . These observations strongly indicate that the  $\text{MoS}_2$  flakes were not noticeably etched during  $\text{BCl}_3$  plasma-derived radical treatment. In the corresponding PL spectra of the monolayer  $\text{MoS}_2$  flakes before and after treatment, no evidence of etching or degradation of  $\text{MoS}_2$  was observed, as there were no significant changes, except for a slight increase in the intensity of the main peak. This change in the main PL peak is consistent with previous research on the chemical doping of  $\text{MoS}_2$ ,<sup>21</sup> suggesting that radicals adsorbed on  $\text{MoS}_2$  may induce subtle doping effects.

To further confirm the minimal adverse effects of the  $\text{BCl}_3$  plasma-derived radical treatment on  $\text{MoS}_2$ , bottom-gated FETs were fabricated using a thick bulk  $\text{MoS}_2$  flake (tens of nanometers) and a  $\text{SiO}_2$  gate dielectric (90 nm). The drain current–gate voltage ( $I_D\text{-}V_G$ ) characteristics were measured at a fixed drain voltage of 0.1 V, as shown in Figure 4b. There was no increase in the off-current level of the bottom-gated FET after the  $\text{BCl}_3$  plasma-derived radical treatment, indicating that there was no detectable damage on the  $\text{MoS}_2$  channel. However, a noticeable shift in the  $I_D\text{-}V_G$  curve toward a positive gate voltage was observed after surface treatment. This shift can be attributed to the doping effect caused by the binding of B to  $\text{MoS}_2$ , which is consistent with that observed in the PL spectra (Figure 4a). Furthermore, the increase in the on-current level of the device with the surface treatment could be attributed to the removal of absorbed OH species on the channel, which degraded the electrical properties of the device,<sup>22</sup> because of the  $\text{BCl}_3$  plasma-generated radicals.

To demonstrate the potential application of the  $\text{BCl}_3$  plasma-derived radical treatment on  $\text{MoS}_2$  for subsequent

ALD of high- $k$  films, top-gated FETs were fabricated using a bulk MoS<sub>2</sub> flake and an ultrathin Al<sub>2</sub>O<sub>3</sub> gate dielectric (target thickness of approximately 5 nm on Si). The resulting device structure and representative  $I_D$ - $V_G$  characteristics are shown in Figure 4c. Although the gate dielectric was extremely thin, the gate leakage current was considerably low and stable. This improvement could be attributed to the uniform and pinhole-free deposition of the ultrathin Al<sub>2</sub>O<sub>3</sub> gate dielectric film on the MoS<sub>2</sub> surface, facilitated by BCl<sub>3</sub> plasma-derived radical pretreatment. The distribution of electrical parameters for the top-gated FETs is plotted and compared with that for the pristine bottom-gated FETs in Figure S3. The average values of the extracted parameters for the top-gated FETs were as follows: on/off ratio of  $\sim 10^6$ , subthreshold swing of  $\sim 140$  mV/dec, and field-effect mobility of  $\sim 9$  cm<sup>2</sup>/(V s). These values indicate a reasonably high performance of the top-gated MoS<sub>2</sub>-based FETs, which further highlights the potential of extending the technology to GAAFETs with a 2D channel material.

#### 4. CONCLUSIONS

In conclusion, our study demonstrated the potential of the BCl<sub>3</sub> plasma-derived radical treatment as an effective 3D surface modification technique for MoS<sub>2</sub>. This treatment improved the coverage and conformality of the ALD-Al<sub>2</sub>O<sub>3</sub> films on both planar and suspended MoS<sub>2</sub> surfaces, demonstrating its applicability for future FETs with GAA structures. Our findings, supported by the OES, indicate that BCl<sub>3</sub> gas primarily decomposed into BCl radicals in the plasma state. Based on DFT calculations, the adsorption of these radicals on MoS<sub>2</sub> facilitated the chemical reactions with an Al precursor, enhancing the nucleation of ALD-Al<sub>2</sub>O<sub>3</sub> films on MoS<sub>2</sub> surfaces. The reaction pathways determined through DFT calculations were experimentally verified through XPS analysis of the samples under different process conditions. Furthermore, a comparison of the Raman and PL spectra of monolayer MoS<sub>2</sub> and the electrical properties of bottom-gated bulk MoS<sub>2</sub> FETs before and after the BCl<sub>3</sub> plasma-derived radical treatment proved that the MoS<sub>2</sub> surface remained undamaged with some observed doping effects. Finally, the practicality of the treatment in terms of device performance was demonstrated through the electrical characterization of a top-gated bulk MoS<sub>2</sub> FET with an ultrathin ALD Al<sub>2</sub>O<sub>3</sub> gate dielectric ( $\sim 5$  nm). This study provides valuable insights and experimental evidence for the application of BCl<sub>3</sub> plasma-derived radical treatment in the surface functionalization of MoS<sub>2</sub>, thereby providing possibilities for the development of high-performance 3D-structured FET devices based on 2D materials.

#### ■ ASSOCIATED CONTENT

##### SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsami.3c09311>.

SEM images of ALD Al<sub>2</sub>O<sub>3</sub> on MoS<sub>2</sub> with different process variables for BCl<sub>3</sub> plasma-derived radical treatment, fabrication process scheme of MoS<sub>2</sub> bridge sample, and statistical distribution of electrical parameters of bottom-gated and top-gated bulk MoS<sub>2</sub> FETs (PDF)

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##### Notes

The authors declare no competing financial interest.

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